Storage of Sugar Cane Bagasse

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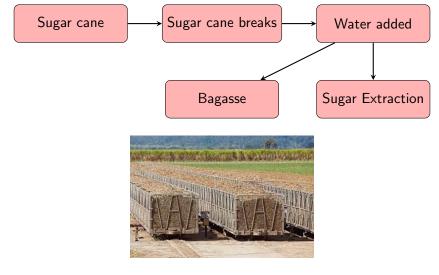


Figure: Moisture levels of 45 - 55%

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Problem Description

- Stockpile as a resource
- Spontaneous combustion
- T. F. Dixon (1988)
- B. F. Gray et al (2002)

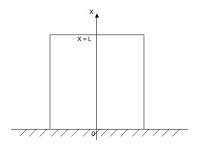


Figure: One Dimensional Model with an insulated bottom

• Maximum height of the bagasse heap to avoid spontaneous combustion?

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- Advantage in adjusting the moisture? (Usable energy per unit area)
- Advantage in pelletizing the bagasse?(Usable energy per unit area)

1D-Model formulation: B. F. Gray et. al 2001

Governing equations

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$$(\rho_b c_b + m_w X c_w) \frac{\partial U}{\partial t} = Q \rho_b Z W \exp(-E/RU) + Q_w \rho_b Z_w X W \exp(-E_w/RU) f(U) + L_v [Z_c Y - Z_e X \exp(-L_v/RU)] + \kappa \nabla^2 U, \quad (1)$$

$$\frac{\partial Y}{\partial t} = Z_e X \exp(-L_v/RU) - Z_c Y + D_Y \nabla^2 Y, \qquad (2)$$
$$\frac{\partial X}{\partial t} = -Z_e X \exp(-L_v/RU) + Z_c Y, \qquad (3)$$

$$\frac{\partial W}{\partial t} = -F\rho_b ZW \exp(-E/RU) - F\rho_b Z_w XW \exp(-E_w/RU)f(U) + D_w \nabla^2 W.$$
(4)

U is temperature, Y is vapour concentration, X is liquid concentration, W is oxygen concentration

1D-Model formulation cont'd

Boundary Conditions

At the bottom, x = 0, we impose the no flow condition (of heat or material)

$$\frac{\partial U}{\partial x} = 0, \qquad \frac{\partial Y}{\partial x} = 0, \qquad \frac{\partial W}{\partial x} = 0,$$
 (5)

At the top surface, x = L,

$$k\frac{\partial U}{\partial x} = h(U - U_a), -D_Y \frac{\partial Y}{\partial x} = h_Y(Y - Y_a), -D_W \frac{\partial W}{\partial x} = h_W(W - W_a),$$
(6)

Initial Conditions

$$U(x,0) = U_0(x), Y(x,0) = Y_0(x), (7)$$

$$X(x,0) = X_0(x), W(x,0) = W_0(x). (8)$$

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1D-Model formulation cont'd

Steady-state equations

$$0 = D_Y \frac{\partial^2 Y}{\partial x^2} + Z_e X \exp\left(-\frac{L_v}{RU}\right) - Z_c Y$$
(9)

$$0 = -Z_e X \exp\left(-\frac{L_v}{RU}\right) + Z_c Y \tag{10}$$

$$Y_{xx} = 0 \Rightarrow Y_s = Y_a, \ X_s = \frac{Z_c Y_a}{Z_e} \exp\left(\frac{L_v}{RU}\right)$$
 (11)

$$0 = k \frac{\partial^2 U}{\partial x^2} + Q \rho_b Z W \exp\left(-\frac{E}{RU}\right) + Q_w \rho_b Z_w X_s W \exp\left(-\frac{E_w}{RU}\right) f(U)$$
(12)

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$$0 = D_W \frac{\partial^2 W}{\partial x^2} - F \rho_b Z W \exp\left(-\frac{E}{RU}\right) - F \rho_b Z_w X W \exp\left(-\frac{E_w}{RU}\right) f(U)$$
(13)

If bagasse is hot (everywhere above 58C), then

$$\frac{k}{Q}\frac{\partial^2 U}{\partial x^2} + \frac{D_W}{F}\frac{\partial^2 W}{\partial x^2} = 0$$
(14)

Applying boundary conditions at x = 0

$$\frac{k}{Q}U + \frac{D_W}{F}W = C_0 \tag{15}$$

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$$\Delta Y = Y_a$$
 $\Delta W = W_a$ $\Delta X = \frac{Z_c Y_a}{Z_e} \exp\left(\frac{L_v}{RU_i}\right)$ $\Delta U = U_i - U_a$

Non-dimensional model

$$\hat{t} = \frac{t}{\Delta t}, \qquad \hat{x} = \frac{x}{L}, \qquad \hat{U} = \frac{U - U_a}{\Delta U}, \qquad \hat{Y} = \frac{Y}{\Delta Y},$$
$$\hat{X} = \frac{X}{\Delta X}, \qquad \hat{W} = \frac{W}{\Delta W}, \qquad (16)$$
$$I^2(abCb + m_W Cw \Delta X) = I^2$$

Diffusion time scale is
$$\Delta t = \frac{L(\rho_b c_b + m_w c_w \Delta \lambda)}{k} = \frac{L}{D_U},$$
 (17)

The liquid equation is

$$\frac{1}{Z_e \Delta t} \exp\left(\frac{L_v}{RU_i}\right) \frac{\partial \hat{X}}{\partial \hat{t}} = -\hat{X} \exp\left(\frac{\alpha_{L_v}(\hat{U}-1)}{U_a + \Delta U\hat{U}}\right) + \hat{Y},$$
(18)

where

$$\alpha_{L_{\nu}} = \frac{L_{\nu} \Delta U}{R U_i}.$$

(19)

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Coefficient of LHS is $\mathcal{O}(10^{-5})$, hence

$$\hat{X} = \exp\left(-\frac{\alpha_{L_{\nu}}(\hat{U}-1)}{U_{a}+\Delta U\hat{U}}\right)\hat{Y}$$
(20)

Lose terms in heat and vapour equations Vapour equation

$$\kappa_{Y} \frac{\partial \hat{Y}}{\partial \hat{t}} = \frac{\partial^{2} \hat{Y}}{\partial \hat{x}^{2}}, \quad \text{where} \quad \kappa_{Y} = \frac{L^{2}}{\Delta t D_{Y}} = \mathcal{O}(10^{-1}). \tag{21}$$
$$(\beta_{1} + \beta_{2} \hat{X}) \frac{\partial \hat{U}}{\partial \hat{t}} = \frac{\partial^{2} \hat{U}}{\partial \hat{x}^{2}} + A_{E} \hat{W} \exp\left(\frac{\alpha_{E}(\hat{U} - 1)}{U_{a} + \Delta U \hat{U}}\right)$$
$$+ A_{E_{w}} \hat{X} \hat{W} \exp\left(\frac{\alpha_{E_{w}}(\hat{U} - 1)}{U_{a} + \Delta U \hat{U}}\right) f(\hat{U}), \tag{22}$$

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where

$$\beta_{1} = \frac{\rho_{b}c_{b}L^{2}}{k\Delta t} \qquad \beta_{2} = \frac{m_{w}c_{w}\Delta XL^{2}}{k\Delta t}$$

$$A_{E} = \frac{Q\rho_{b}Z\Delta WL^{2}}{k\Delta U} \exp\left(-\frac{E}{RU_{i}}\right)$$

$$A_{E_{w}} = \frac{Q_{w}\rho_{b}Z_{w}\Delta X\Delta WL^{2}}{k\Delta U} \exp\left(-\frac{E_{w}}{RU_{i}}\right),$$

$$\alpha_{E} = \frac{E\Delta U}{RU_{i}}, \qquad \alpha_{E_{w}} = \frac{E_{w}\Delta U}{RU_{i}}$$
(23)
(24)
(25)
(25)

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The oxygen equation becomes

$$\kappa_{W} \frac{\partial \hat{W}}{\partial \hat{t}} = \frac{\partial^{2} \hat{W}}{\partial \hat{x}^{2}} - B_{E} \hat{W} \exp\left(\frac{\alpha_{E}(\hat{U}-1)}{U_{a} + \Delta U\hat{U}}\right) - B_{E_{w}} \hat{X} \hat{W} \exp\left(\frac{\alpha_{E_{w}}(\hat{U}-1)}{U_{a} + \Delta U\hat{U}}\right) f(\hat{U}),$$
(27)

where

$$\kappa_{W} = \frac{L^{2}}{\Delta t D_{W}} \quad B_{E} = \frac{F \rho_{b} Z L^{2}}{D_{W}} \exp\left(-\frac{E}{R U_{i}}\right)$$

$$B_{E_{w}} = \frac{F \rho_{b} Z_{w} \Delta X L^{2}}{D_{w}} \exp\left(-\frac{E_{w}}{R U_{i}}\right).$$
(28)
(29)

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Boundary conditions

At
$$\hat{x} = 0$$
: $\frac{\partial \hat{U}}{\partial \hat{x}} = 0$, $\frac{\partial \hat{Y}}{\partial \hat{x}} = 0$, $\frac{\partial \hat{W}}{\partial \hat{x}} = 0$, at $\hat{x} = 0$,
(30)
At $\hat{x} = 1$:
 $-\frac{\partial \hat{U}}{\partial \hat{x}} = \gamma \hat{U}$, $-\frac{\partial \hat{Y}}{\partial \hat{x}} = \gamma_{Y}(\hat{Y} - 1)$, $-\frac{\partial \hat{W}}{\partial \hat{x}} = \gamma_{W}(\hat{W} - 1)$,
(31)

where

$$\gamma = \frac{hL}{k}, \qquad \gamma_Y = \frac{h_Y L}{D_Y}, \qquad \gamma_W = \frac{h_W L}{D_W}.$$
 (32)

Note $\gamma = \mathcal{O}(10)$, $\gamma_Y = \gamma_W = \mathcal{O}(10^5)$ so we may simplify the boundary conditions $\hat{Y} = \hat{W} = 1$ at $\hat{x} = 1$. The initial conditions are

$$U = U_0, \qquad Y = Y_0, \qquad W = W_0, \quad \text{at } t = 0$$
 (33)

Simplest model

Steady-state temperature

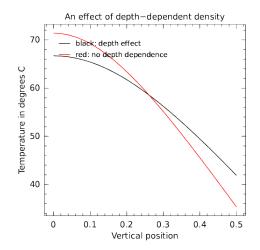
$$0 = \frac{\partial^2 \hat{U}}{\partial \hat{x}^2} + A \exp\left(\frac{\alpha(\hat{U} - 1)}{U_a + \Delta U\hat{U}}\right)$$
(34)

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This is standard form, but leads to very small piles

Discussion

What happens when the density is not assumed constant?



Discussion

Pseudo steady-state

 κ_W, κ_Y small

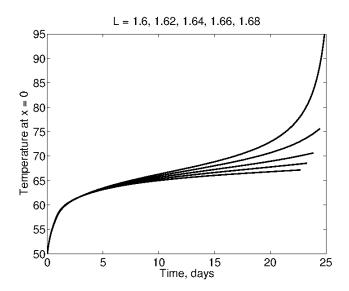
$$\hat{X} = \exp\left(-\frac{\alpha_{L_{\nu}}(\hat{U}-1)}{U_{a}+\Delta U\hat{U}}\right)\hat{Y} \qquad \hat{Y} = 1$$
(35)
$$(\beta_{1}+\beta_{2}\hat{X})\frac{\partial\hat{U}}{\partial\hat{t}} = \frac{\partial^{2}\hat{U}}{\partial\hat{x}^{2}} + A_{E}\hat{W}\exp\left(\frac{\alpha_{E}(\hat{U}-1)}{U_{a}+\Delta U\hat{U}}\right) \\
+ A_{E_{w}}\hat{X}\hat{W}\exp\left(\frac{\alpha_{E_{w}}(\hat{U}-1)}{U_{a}+\Delta U\hat{U}}\right)f(\hat{U}),$$
(36)
$$0 = \frac{\partial^{2}\hat{W}}{\partial\hat{x}^{2}} - B_{E}\hat{W}\exp\left(\frac{\alpha_{E}(\hat{U}-1)}{U_{a}+\Delta U\hat{U}}\right) \\
- B_{E_{w}}\hat{X}\hat{W}\exp\left(\frac{\alpha_{E_{w}}(\hat{U}-1)}{U_{a}+\Delta U\hat{U}}\right)f(\hat{U}),$$
(37)

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Discussion

Almost full problem

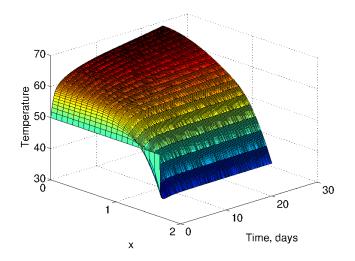
$$\hat{X} = \exp\left(-\frac{\alpha_{L_{v}}(\hat{U}-1)}{U_{a}+\Delta U\hat{U}}\right)\hat{Y}$$
(38)
$$\kappa_{Y}\frac{\partial\hat{Y}}{\partial\hat{t}} = \frac{\partial^{2}\hat{Y}}{\partial\hat{x}^{2}},$$
(39)
$$(\beta_{1}+\beta_{2}\hat{X})\frac{\partial\hat{U}}{\partial\hat{t}} = \frac{\partial^{2}\hat{U}}{\partial\hat{x}^{2}} + A_{E}\hat{W}\exp\left(\frac{\alpha_{E}(\hat{U}-1)}{U_{a}+\Delta U\hat{U}}\right) + A_{E_{w}}\hat{X}\hat{W}\exp\left(\frac{\alpha_{E_{w}}(\hat{U}-1)}{U_{a}+\Delta U\hat{U}}\right)f(\hat{U}),$$
(40)
$$\kappa_{W}\frac{\partial\hat{W}}{\partial\hat{t}} = \frac{\partial^{2}\hat{W}}{\partial\hat{x}^{2}} - B_{E}\hat{W}\exp\left(\frac{\alpha_{E}(\hat{U}-1)}{U_{a}+\Delta U\hat{U}}\right) - B_{E_{w}}\hat{X}\hat{W}\exp\left(\frac{\alpha_{E_{w}}(\hat{U}-1)}{U_{a}+\Delta U\hat{U}}\right)f(\hat{U}),$$
(41)



Note, insulated bottom and 100% humidity. Pile height increases with lower humidity

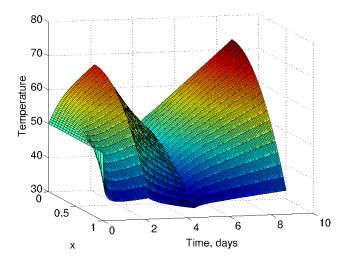
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Typical evolution of temperature

Often appears piles can be very large without ignition but ...



What if it rains?

Puzzle: why do apparently stable heaps ignite after getting soaked?

— wet reaction is fast, but turns off for temperatures above 58 $^\circ\mathrm{C},$ — dry reaction is slower

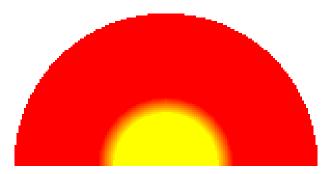
So, near centre the bagasse dries out and starts to heat above the 58 $^\circ \rm C$ limit. We imagine two steady states:

inner: hot and dry, insulated inner end, at 58 $^\circ\text{C}$ at interface outer: warm and wet, 58 $^\circ\text{C}$ at interface, cooling condition at surface

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Stefan problem with moving boundary

Ignition model — diagram



Please use conformal mapping to imagine this as a square with a hot yellow and a warm red band ...

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We have a good handle on equations for the steady states, but haven't got a formulation for the velocity of the moving interface

• We have a model for temperature evolution in bagasse piles - can be made simpler

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• Steady-state models should be sufficient - to provide bifurcation diagram

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• Under normal conditions pile does not burn, but adding water can then cause ignition

- We have a model for temperature evolution in bagasse piles can be made simpler
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- For any ambient conditions we can cause ignition, by making the pile sufficiently large
- Under normal conditions pile does not burn, but adding water can then cause ignition
- We have looked at a worst case scenario insulated bottom, no heat loss at sides. Model can be improved.
- Future work will constitute consideration of a more realistic boundary condition at the bottom, 2D model with heat loss at the sides; compare full system to simplified models.

- B. F. Gray, M. J. Sexton, B. Halliburton, C. Macaskill. Wetting-induced ignition in cellulosic materials. Fire Safety J 37 (2002) 465 - 479.
- T. F. Dixon. Spontaneous combustion in bagasse stockpiles. Proceedingsof the Australian Sugar Cane Technology, Mackay, Queensland, Australian, 1988, p. 53 - 61.
- O. Macaskill, M. J. Sexton, B. F. Gray. A reaction diffusion model of stored bagasse. Anziam J 43 (2001) 13 - 35.

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